

96. Some Observations Concerning $^1\text{H-NMR}$. Spectra of 5-Hydroxy- and 5-Acetoxy-steroid Compounds

by Mihailo Lj. Mihailović, Ljubinka Lorenc and Vladimir Pavlović

Department of Chemistry, Faculty of Science, University of Belgrade,
and Institute of Chemistry, Technology and Metallurgy, Studentski trg 16, POB 550, YU-11001 Belgrade,
Yugoslavia

and Hermann Fuhrer

Central Function Research, Ciba-Geigy Ltd., CH-4002 Basle

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Summary

It was found that analysis of the $^1\text{H-NMR}$. signals (at 360 (preferably) and/or 100 MHz) of the protons at C(4) and of the equatorial α -proton at C(6), particularly when these are located in the 2–3 ppm region and therefore convenient for detection and identification, may be of valuable aid in the structural and configurational characterization of 5-hydroxy- and 5-acetoxy-steroids (unsubstituted or containing a hydroxy or acyloxy substituent at C(3)).

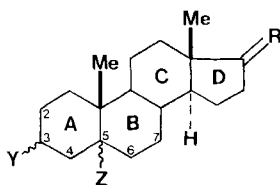
Introduction. – Although $^1\text{H-NMR}$. spectroscopy has been extensively applied to steroid compounds [1–3], little attention has been paid, so far, to the relationship between substitution at C(5) (and C(3)) and the position of the resonance signals of protons at adjacent C-atoms, *i.e.* at C(4) and C(6). For that reason, we wish to present in this paper some observations on the $^1\text{H-NMR}$. chemical shifts (at 360 (preferably) and/or 100 MHz) of the equatorial α -proton at C(6) and of both protons at C(4) of 5 α - and 5 β -steroids containing a hydroxy- or acetoxy-substituent at C(5), and without or with a substituent (hydroxy, acetoxy, *p*-nitrobenzoyloxy group) at C(3). The structural types used in this study are shown in the *Scheme* and the relevant $^1\text{H-NMR}$. data are given in the *Table*¹.

Discussion. – 1. *Equatorial α -proton at C(6).* In 5 α - and 5 β -unsubstituted steroids **1** and **2** (A or B, Z=H), respectively, even in those containing a 3 α - or 3 β -*p*-nitrobenzoyloxy group (Y=PNBO) [2], the signal of the equatorial $\text{H}_\alpha\text{-C}(6)$ (not shown in the *Table*) is located at or upfield of 2.0 ppm, and is not easily identifiable, particularly in 60- and 100-MHz- $^1\text{H-NMR}$. spectra.

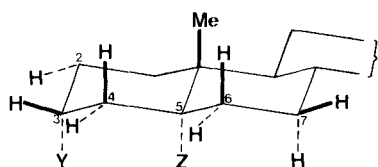
The same is true for 5 α - and 5 β -hydroxy-steroids **1** and **2** (Z=HO), respectively, in which the resonance of the equatorial $\text{H}_\alpha\text{-C}(6)$ appears at 1.97–2.03 ppm,

¹) Chemical shifts are given in ppm relative to tetramethylsilane ($\delta = 0$ ppm) as internal standard.

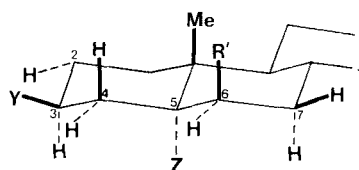
Scheme



1 5 α -Z; 2 5 β -Z
 A 3 α -Y; B 3 β -Y

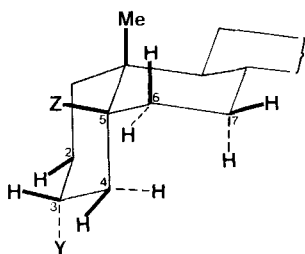


1A 3 α -Y_{ax}, 5 α -Z_{ax} (3 β -H_{eq})

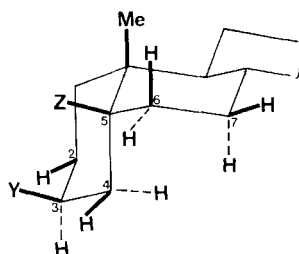


1B (R'=H), 1BB (R'=AcO)
 3 β -Y_{eq}, 5 α -Z_{ax} (3 α -H_{ax})

A/B = *trans*
 5 α -steroids 1



2A 3 α -Y_{eq}, 5 β -Z_{ax}.eq (3 β -H_{ax})



2B 3 β -Y_{ax}, 5 β -Z_{ax}.eq (3 α -H_{eq})

A/B = *cis*
 5 β -steroids 2

aa Z=HO; Y=H; dd Z=AcO, Y=H;
 a Z=Y=HO; b Z=HO, Y=AcO; c Z=HO, Y=PNBO; d Z=AcO, Y=HO; e Z=Y=AcO;
 f Z=AcO, Y=PNBO; g Z=CCl₃CONHCOO, Y=AcO
 (PNBO = *p*-NO₂C₆H₄COO)

regardless of substitution at C(3) (Y = H, α - or β -HO, -AcO, -PNBO). (The value of 2.03 ppm is observed for 3 β , 5 β -cholestanediol 3-*p*-nitrobenzoate (**2Bc**)).

However, when the substituent at C(5) is an acetoxy group (Z = AcO) or *N*-trichloroacetyl-carbamoyloxy group (Z = CCl₃CONHCOO), its ester carbonyl moiety assumes such a conformation as to cause deshielding of the equatorial H_a-C(6), both in the 5 α - and 5 β -series. The signal of this H_a-C(6) (irrespective of substitution at C(3), *i.e.* Y = H, α - or β -HO, -AcO, -PNBO) is now readily detectable around 2.5 ppm, *i.e.* at 2.44–2.55 ppm for the 5 α -acetoxy compounds (**1dd**; **1Ad**, **1Ae**, **1Af**; **1Bd**, **1Be**, **1Bf**), and at 2.48–2.62 ppm for the 5 β -acetoxy steroids (**2dd**; **2Ad**, **2Ae**, **2Af**; **2Be**, **2Bf**, **2Bg**)²⁾.

2. *Protons at C(4)*. In 5 α - and 5 β -unsubstituted steroids **1** and **2** (Z = H), respectively, without or with substituents at C(3) (Y = H, α - or β -HO, -AcO, -PNBO) [2], the two protons at C(4) resonate upfield of 2.0 ppm. An exception is 5 β -cholestan-3 β -ol *p*-nitrobenzoate (**2B**, Z = H, Y = PNBO) [2], which shows a

Table. Chemical shifts of protons at C(4) and C(6) in various 5-hydroxy- and 3,5-dihydroxy-steroids and their acetate and *p*-nitrobenzoate esters **1** and **2** (see Scheme^a)

5 α -Steroid ^{b)}	MHz	¹ H-NMR. signals (ppm) ^{c)}		
		H _a -C(4) ^{d)} equatorial	H _{β} -C(4) ^{e)} axial	H _a -C(6) ^{f)} equatorial
5 α -Z _{ax} (1)				
1aa Z = HO Y = H	360	g)	g)	1.98 $d \times t$ ($J_{gem} = 12$, $J_{vic} = 4$)
1dd Z = AcO Y = H	100	2.45 ^{h)} m ($J_{gem} \approx 12-14$)	g)	2.45 ^{h)} m ($J_{gem} \approx 12-14$)
3 α -Y _{ax} , 5 α -Z _{ax} (1A) ⁱ⁾				
1Aa Z = HO Y = HO	360	1.61 $d \times d \times d$ ($J_{gem} = 16$, $J_{vic} = 2.5$, $J_{2eq,4eq} = 2$)	1.79 $d \times d$ ($J_{gem} = 16$, $J_{vic} = 4$)	1.98 $d \times t$ ($J_{gem} = 12$, $J_{vic} = 4$)
1Aa irradi. at 4.07 ppm (m , $w_{1/2} \approx 7$, = eq. H _{β} -C(3))		1.61 $d \times d$ ($J_{gem} = 16$, $J_{2eq,4eq} = 2$)	1.79 d ($J_{gem} = 16$)	1.98 $d \times t$ ($J_{gem} = 12$, $J_{vic} = 4$)
1Ab Z = HO Y = AcO	360	1.64 $d \times t$ ($J_{gem} = 16$, $J_{vic} = J_{2eq,4eq} = 2$)	1.89 $d \times d$ ($J_{gem} = 16$, $J_{vic} = 4$)	1.99 $d \times t$ ($J_{gem} = 12$, $J_{vic} = 4$)
1Ab irradi. at 5.20 ppm (m , $w_{1/2} \approx 8$, = eq. H _{β} -C(3))		1.64 $d \times d$ ($J_{gem} = 16$, $J_{2eq,4eq} = 2$)	1.89 d ($J_{gem} = 16$)	1.99 $d \times t$ ($J_{gem} = 12$, $J_{vic} = 4$)
1Ad Z = AcO Y = HO	100	2.82 $d \times t$ ($J_{gem} = 16$, $J_{vic} = J_{2eq,4eq} = 2$)	g)	2.49 $d \times t$ ($J_{gem} = 14$, $J_{vic} = 3$)

2) For the reason already discussed previously [3] [4], the signal of the equatorial H_a-C(6) in the steroidal 3 β , 5 α , 6 β -triol 3,6-diacetates **1BBb** and **1BBb'**, which is situated at 4.75–4.76 ppm, is shifted downfield by about 1.2 ppm when the 5 α -hydroxy group is acetylated, appearing at 5.92–5.93 ppm in the triacetates **1BBe** and **1BBe'** (see Table).

Table (continued)

5 α -Steroid ^{b)}	MHz	¹ H-NMR. signals (ppm) ^{c)}		
		H $_{\alpha}$ -C(4) ^{d)} equatorial	H $_{\beta}$ -C(4) ^{e)} axial	H $_{\alpha}$ -C(6) ^{f)} equatorial
1Ae Z = AcO Y = AcO	360	2.84 $d \times t$ ($J_{\text{gem}} = 16$, $J_{\text{vic}} = J_{2\text{eq},4\text{eq}} = 2$)	g)	2.44 $d \times t$ ($J_{\text{gem}} = 14$, $J_{\text{vic}} = 3.5$)
1Af Z = AcO Y = PNBOj)	100	3.21 $d \times d \times d$ ($J_{\text{gem}} = 16$, $J_{\text{vic}} = 2.5$, $J_{2\text{eq},4\text{eq}} = 2$)	g)	2.44 $d \times t$ ($J_{\text{gem}} = 14$, $J_{\text{vic}} = 3$)
3 β -Y $_{\text{eq}}$ 5 α -Z $_{\text{ax}}$ (1B) ^{k)}				
1Ba Z = HO Y = HO	360	g)	g)	1.99 $d \times t$ ($J_{\text{gem}} = 12$, $J_{\text{vic}} = 4$)
1Bb Z = HO Y = AcO	360	g)	g)	1.98 $d \times t$ ($J_{\text{gem}} = 12$, $J_{\text{vic}} = 4$)
1Bb irradi. at 5.16 ppm (m , $w_{1/2} \approx 23$, = ax. H $_{\alpha}$ -C(3))		g)	g)	1.98 $d \times t$ ($J_{\text{gem}} = 12$, $J_{\text{vic}} = 4$)
1Bc Z = HO Y = PNBOj)	360	g)	g)	2.00 $d \times t + m$ (1H) ($J_{\text{gem}} = 12$, $J_{\text{vic}} = 3$)
1Bc irradi. at 5.47 ppm (sept, l.s. ≈ 5 , = ax. H $_{\alpha}$ -C(3))		g)	g)	ditto
1Bd Z = AcO Y = HO	100	2.83 $d \times d \times d$ ($J_{\text{gem}} = 14$, $J_{\text{vic}} = 5$, $J_{2\text{eq},4\text{eq}} = 2$)	g)	2.52 $d \times t$ ($J_{\text{gem}} = 14$, $J_{\text{vic}} = 3$)
1Be Z = AcO Y = AcO	100	2.80 $d \times d \times d$ ($J_{\text{gem}} = 14$, $J_{\text{vic}} = 5$, $J_{2\text{eq},4\text{eq}} = 2$)	g)	2.54 $d \times t$ ($J_{\text{gem}} = 14$, $J_{\text{vic}} = 3$)
1Bf Z = AcO Y = PNBOj)	360	2.96 $d \times d \times d$ ($J_{\text{gem}} = 16$, $J_{\text{vic}} = 5$, $J_{2\text{eq},4\text{eq}} = 2$)	g)	2.55 $d \times t$ ($J_{\text{gem}} = 14$, $J_{\text{vic}} = 3$)
1Bf irradi. at 5.07 ppm (sept, l.s. ≈ 5 , = ax. H $_{\alpha}$ -C(3))		2.96 $d \times d$ ($J_{\text{gem}} = 16$, $J_{2\text{eq},4\text{eq}} = 2$)	g)	2.55 $d \times t$ ($J_{\text{gem}} = 14$, $J_{\text{vic}} = 3$)
1BBb ^{l)} Z = HO Y = AcO	60	g)	g)	4.75 m ($w_{1/2} \approx 5$)
1BBb ^{l)} Z = HO Y = AcO R = H $_2$	60	g)	g)	4.76 m ($w_{1/2} \approx 5$)
1BBe ^{l)} Z = AcO Y = AcO	60	2.85 $d \times d \times d$ ($J_{\text{gem}} = 14$, $J_{\text{vic}} = 5$, $J_{2\text{eq},4\text{eq}} = 2$)	g)	5.92 m ($w_{1/2} \approx 5.5$)
1BBe ^{l)} Z = AcO Y = AcO R = H $_2$	60	2.87 $d \times d \times d$ ($J_{\text{gem}} = 14$, $J_{\text{vic}} = 5$, $J_{2\text{eq},4\text{eq}} = 2$)	g)	5.93 m ($w_{1/2} \approx 5$)

Table (continued)

5 β -Steroid ^{b)}	MHz	¹ H-NMR. signals (ppm) ^{c)}		
		H _{α} -C(4) ^{m)} axial	H _{β} -C(4) ⁿ⁾ equatorial	H _{α} -C(6) ^{f)} equatorial
5 β -Z _{ax,eq} (2) ^{o)}				
2aa Z = HO Y = H	360	2.13 $d \times d \times d$ ($J_{gem} = 14, J_{vic} = 12,$ $J_{vic} = 6$)	g)	1.98 $d \times t$ ($J_{gem} = 13, J_{vic} = 3$)
2dd Z = AcO Y = H		g)	2.51 ^{h)} m ($J_{gem} \approx 14$)	2.51 ^{h)} m ($J_{gem} \approx 14$)
3 α -Y _{eq} , 5 β -Z _{ax,eq} (2A) ^{k)} ^{o)}				
2Aa Z = HO Y = HO	360	2.03 ^{h)} m ($J_{gem} \approx 14$)	g)	1.98 ^{h)} m ($J_{gem} \approx 14$)
2Ab Z = HO Y = AcO	100	2.15 $d \times d$ ($J_{gem} = 14, J_{vic} = 12$)	g)	g)
2Ad Z = AcO Y = HO	360	g)	2.81 $d \times d \times d$ ($J_{gem} = 14, J_{vic} = 5,$ $J_{2eq,4eq} = 2$)	2.54 $d \times d \times d$ ($J_{gem} = 13, J_{vic} = 4,$ $J_{vic} = 3$)
2Ad irradi. at 3.73 ppm (sept, i.s. $\approx 5,$ = ax. H _{β} -C(3))		g)	2.81 $d \times d$ ($J_{gem} = 14, J_{2eq,4eq} = 2$)	2.54 $d \times d \times d$ ($J_{gem} = 13, J_{vic} = 4,$ $J_{vic} = 3$)
2Ae Z = AcO Y = AcO	360	g)	2.78 $d \times d \times d$ ($J_{gem} = 14, J_{vic} = 5,$ $J_{2eq,4eq} = 2$)	2.58 $d \times d \times d$ ($J_{gem} = 14, J_{vic} = 5,$ $J_{vic} = 3$)
2Af Z = AcO Y = PNBOj)	360	2.22 $d \times d$ ($J_{gem} = 14, J_{vic} = 12$)	2.98 $d \times d \times d$ ($J_{gem} = 14, J_{vic} = 5,$ $J_{2eq,4eq} = 2$)	2.62 $d \times t$ ($J_{gem} = 14, J_{vic} = 4$)
3 β -Y _{ax} , 5 β -Z _{ax,eq} (2B) ⁱ⁾ ^{o)}				
2Ba Z = HO Y = HO	100	2.23 $d \times d$ ($J_{gem} = 15, J_{vic} = 3.5$)	g)	g)
2Ba' Z = HO Y = HO R = H ₂	100	2.22 $d \times d$ ($J_{gem} = 15, J_{vic} = 3.5$)	g)	g)
2Ba'' Z = HO Y = HO R = β -HO, α -H	100	2.22 $d \times d$ ($J_{gem} = 15, J_{vic} = 4$)	g)	g)
2Ba''' Z = HO Y = HO R = O	360	2.18 ^{p)} $d \times d$ ($J_{gem} = 15, J_{vic} = 3.5$)	g)	g)
2Ba''' irradi. at 4.17 ppm ($m, w_{1/2} \approx 7,$ = eq. H _{α} -C(3))	100	2.18 d ($J_{gem} = 15$)	g)	g)
2Bb Z = HO Y = AcO	100	2.32 $d \times d$ ($J_{gem} = 16, J_{vic} = 4$)	g)	g)
2Bb' Z = HO Y = AcO R = H ₂	100	2.34 $d \times d$ ($J_{gem} = 16, J_{vic} = 4$)	g)	g)

Table (continued)

5 β -Steroid ^{b)}	MHz	¹ H-NMR. signals (ppm) ^{c)}		
		H _{α} -C(4) ^{m)} axial	H _{β} -C(4) ⁿ⁾ equatorial	H _{α} -C(6) ^{l)} equatorial
2Bb' Z = HO Y = AcO R = β -AcO, α -H	100	2.28 $d \times d$ ($J_{\text{gem}} = 16, J_{\text{vic}} = 4$)	g)	g)
2Bb'' + TAI ^{q)} (= 2Bg'') Z = CCl ₃ CONHCOO Y = AcO R = β -AcO, α -H	100	2.20 $d \times d$ ($J_{\text{gem}} = 16, J_{\text{vic}} = 4$)	2.92 $d \times t$ ($J_{\text{gem}} = 16, J_{\text{vic}} \approx 2$)	2.52 $d \times t$ ($J_{\text{gem}} = 14, J_{\text{vic}} = 3$)
2Bc Z = HO Y = PNBO ⁱ⁾	360	2.43 $d \times d$ ($J_{\text{gem}} = 16, J_{\text{vic}} = 4$)	g)	2.03 $d \times t$ ($J_{\text{gem}} = 13, J_{\text{vic}} = 3$)
2Bc irradi. at 5.49 ppm ($m, w_{1/2} \approx 7,$ = eq. H _{α} -C(3))		2.43 d ($J_{\text{gem}} = 16$)	g)	2.03 $d \times t$ ($J_{\text{gem}} = 13, J_{\text{vic}} = 3$)
2Be Z = AcO Y = AcO	100	g)	2.92 $d \times t$ ($J_{\text{gem}} = 16, J_{\text{vic}} \approx 2$)	2.50 $d \times t$ ($J_{\text{gem}} = 13, J_{\text{vic}} = 4$)
2Bf Z = AcO Y = PNBO ⁱ⁾	360	2.21 $d \times d$ ($J_{\text{gem}} = 16.5, J_{\text{vic}} = 4$)	3.18 $d \times t$ ($J_{\text{gem}} = 16.5, J_{\text{vic}} \approx 2$)	2.48 $d \times d \times d$ ($J_{\text{gem}} = 13, J_{\text{vic}} = 5,$ $J_{\text{vic}} = 3$) ^{f)}

- a) ¹H-NMR. signals of other protons in these compounds were given and discussed previously [2] [3].
- b) If not stated otherwise, compounds **1** and **2** (of the A and B series) listed in this Table are cholestane derivatives (R = β -CHMe(CH₂)₃CHMe₂, α -H).
- c) Coupling constants J in Hz. Internal standard: tetramethylsilane (= 0 ppm). Abbreviations: d doublet, t triplet, qa quadruplet, qi quintuplet, $sept.$ septuplet, m multiplet, l.s. line spacing (splitting) in Hz, $w_{1/2}$ half-band width in Hz; gem geminal, vic vicinal, eq equatorial, ax axial.
- d) J_{vic} for eq. H _{α} -C(4) corresponds in compounds **1** to $J_{3\text{ax},4\text{eq}}$ and $J_{3\text{eq},4\text{eq}}$, in **1A** to $J_{3\text{eq},4\text{eq}}$, and in **1B** to $J_{3\text{ax},4\text{eq}}$; $J_{2\text{eq},4\text{eq}}$ refers to 'W'-type-4 σ -bond-long-range coupling.
- e) J_{vic} for ax. H _{β} -C(4) corresponds in compounds **1** to $J_{3\text{ax},4\text{ax}}$ and $J_{3\text{eq},4\text{ax}}$, in **1A** to $J_{3\text{eq},4\text{ax}}$, and in **1B** to $J_{3\text{ax},4\text{ax}}$.
- f) J_{vic} for eq. H _{α} -C(6) in all compounds **1** and **2** corresponds to $J_{6\text{eq},7\text{ax}}$ and $J_{6\text{eq},7\text{eq}}$.
- g) Not easily observable since within the methylene envelope, in the region upfield of 2.0 ppm.
- h) Signals of H-C(4) and eq. H _{α} -C(6) overlapping.
- i) The signal of the eq. H-C(3) in compounds **1A** (i.e. eq. H _{β} -C(3)) and **2B** (i.e. eq. H _{α} -C(3)) appears as a m (ill-resolved qi) with $w_{1/2} \approx 5-8$ Hz.
- j) PNBO = *p*-nitrobenzoyloxy group.
- k) The signal of the ax. H-C(3) in compounds **1B** (i.e. ax. H _{α} -C(3)) and **2A** (i.e. ax. H _{β} -C(3)) appears usually in the 360 MHz spectra as a $sept.$ with l.s. of about 4.5-5.5 Hz, and in 100 MHz spectra as a broad m with $w_{1/2} \approx 22-24$ Hz.
- l) ¹H-NMR. spectra of these compounds were discussed previously [3] [4].
- m) J_{vic} for ax. H _{α} -C(4) corresponds in compounds **2** to $J_{3\text{ax},4\text{ax}}$ and $J_{3\text{eq},4\text{ax}}$, in **2A** to $J_{3\text{ax},4\text{ax}}$, and in **2B** to $J_{3\text{eq},4\text{ax}}$.
- n) J_{vic} for eq. H _{β} -C(4) corresponds in compounds **2** to $J_{3\text{ax},4\text{eq}}$ and $J_{3\text{eq},4\text{eq}}$, in **2A** to $J_{3\text{ax},4\text{eq}}$, and in **2B** to $J_{3\text{eq},4\text{eq}}$; $J_{2\text{eq},4\text{eq}}$ refers to 'W'-type-4 σ -bond-long-range coupling.
- o) Z_{ax,eq} means axial on ring A and equatorial on ring B.
- p) The two four line signals at 2.07 and 2.46 ppm (at 360 MHz) belong to H₂C(16).
- q) TAI = trichloroacetyl-isocyanate (CCl₃CO-N=C=O).
- r) $J_{\text{vic}} = 5$ refers to $J_{6\text{eq},7\text{ax}}$, and $J_{\text{vic}} = 3$ refers to $J_{6\text{eq},7\text{eq}}$.

signal at 2.10 ppm (not given in the *Table*), attributed to the axial $H_a-C(4)$, because, upon irradiation of the equatorial $H_a-C(3)$ at 5.38 ppm, the triplet of doublets centered at 2.10 ppm ($J_{gem} \approx J_{4ax,5ax} \approx 14$ Hz, $J_{3eq,4ax} \approx 3$ Hz) is reduced to a triplet ($J \approx 14$ Hz).

2.1. *Equatorial H-C(4)*. The equatorial proton at C(4) is α in 5 α -steroids **1** and β in 5 β -compounds **2**. Regardless of substitution at C(3) (Y = H, α - or β -HO, -AcO, -PNBO), the signal of the equatorial H-C(4) is located upfield of 2.0 ppm in 5 α - and 5 β -hydroxy-steroids **1** and **2** (Z = HO), respectively (e.g. **1Aa** (Y = HO) and **1Ab** (Y = AcO) in the *Table*).

However, when the 5-hydroxy group is acylated, as in the 5 α - and 5 β -acetoxy-steroids **1** and **2** (**A** and **B**, Z = AcO), respectively, and also in the 5 β -*N*-trichloroacetyl-carbamoyloxy-steroid **2Bg''**, then the equatorial H-C(4) is deshielded by the 5 α - or 5 β -ester carbonyl group (cf. section 1). It now resonates at 2.5-3.2 ppm, the downfield shift increasing somewhat with the deshielding influence (and polarity) of the substituent at C(3) (in order of compound type: **dd** < **d** \approx **e** (\approx **g**) < **f**, in both the 5 α -series **1A** and **1B**, and 5 β -series **2A** and **2B**). Thus, for Y = H the equatorial H-C(4) appears at 2.45-2.5 ppm, for Y = HO or AcO at 2.8-2.9 ppm, and for Y = PNBO at 2.95-3.2 ppm.

It should be noted that when the resonances of both the equatorial $H_a-C(6)$ and the equatorial H-C(4) are situated around 2-3 ppm, the signal of H-C(4) is always at lower field, the difference being in general 0.2-0.4 ppm (but 0.77 and 0.70 ppm for the 5 α - and 5 β -acetoxy compounds containing an axial 3-*p*-nitrobenzoyloxy group. *i.e.* **1Af** and **2Bf**, respectively).

2.2. *Axial H-C(4)*. The axial proton at C(4) is β in 5 α -steroids **1** and α in 5 β -compounds **2**. In the 5 α -series **1**, irrespective of substitution at C(5) (Z = H, HO or AcO) and at C(3) (Y = H, α - or β -HO, -AcO, -PNBO), the signal of this axial $H_\beta-C(4)$ is located upfield of 2.0 ppm (e.g. **1Aa** and **1Ab** in the *Table*), and thus not easily identifiable.

In the 5 β -series **2**, the 5 β -hydroxy-steroids without or with substituents at C(3) (Z = HO; Y = H, α - or β -HO, -AcO, -PNBO) show a signal for the axial $H_a-C(4)$ around 2.0-2.4 ppm, the downfield shift being more pronounced (by about 0.2 ppm) in 5 β -hydroxy compounds with an axial 3 β -substituent (compare **2Aa** and **2Ab** with **2Ba** and **2Bb**).

In 5 β -acetoxy-steroids **2** (Z = AcO; Y = H, α - or β -HO, and α - or β -AcO) the signal of the axial $H_a-C(4)$ is situated upfield of 2.0 ppm, except when the 5 β -acetoxy compound contains an equatorial 3 α - or axial 3 β -*p*-nitrobenzoyloxy group (**2Af** and **2Bf**, respectively), in which case the signal of the axial $H_a-C(4)$ is located at about 2.2 ppm. When the 5 β -acetoxy group is replaced by the 5 β -*N*-trichloroacetyl-carbamoyloxy group, as in **2Bg''**, the axial $H_a-C(4)$ resonates at 2.20 ppm too.

In conclusion it can be said that the appearance of 1H -NMR. signals in the 2-3 ppm region and their assignment to the proton(s) at C(4) and/or the equatorial α -proton at C(6) may be of diagnostic value (in combination with other 1H -NMR. data) for an easier structural and configurational identification and/or characterization of 5-hydroxy- and 5-acetoxy-steroid compounds (unsubstituted or containing a

hydroxy, acetoxy or *p*-nitrobenzoyloxy group at C(3)), particularly when 360-MHz-¹H-NMR. spectra are available.

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Experimental Part

¹H-NMR. spectra were measured in CDCl₃ at 360 MHz on a *Bruker HX-360* spectrometer and/or at 100 MHz on a *Varian XL-100-15* spectrometer, both equipped with a *Fourier transform* accessory. Noise decoupled ¹³C-NMR. spectra were recorded (in CDCl₃) at 25.2 MHz on the same *Varian* apparatus. Deuterium atoms of the solvent (CDCl₃) were used for the 15.4 MHz ²H-lock during ¹H-NMR. and ¹³C-NMR. work. All the spectra were taken at room temperature.

The steroids used in this study were all known compounds, the preparation and properties of which were described previously [2] [3].

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